Product Information



COOCH₂

Leukotriene A₄-d₅ methyl ester

Item No. 10006197

Formal Name: 5S-trans-5,6-oxido-7E,9E,11Z,14Z-

eicosatetraenoic-19,19,20,20, and 20-d₅

acid, methyl ester

Synonym: LTA_4 -d₅ methyl ester

MF: $C_{21}H_{27}D_5O_3$

FW: 337.5

Chemical Purity: ≥97% Leukotriene A₄

Deuterium

 \geq 99% deuterated forms (d₁-d₅); \leq 1% d₀ Incorporation:

Stability: ≥1 year at -80°C

Supplied as: A solution in hexane containing 1%

triethylamine

λ_{max}: 279 nm ε: 49,000 UV/Vis.:

Miscellaneous: Light Sensitive

Laboratory Procedures

Leukotriene A₄-d₅ methyl ester (LTA₄-d₅ methyl ester) contains five deuterium atoms at the 19, 19', 20, 20, and 20 positions. It is intended for use as an internal standard for the quantification of LTA4 methyl ester by GC- or LC-mass spectrometry (MS). For long term storage, we suggest that LTA₄-d₅ methyl ester be stored as supplied at -80°C. It should be stable for at least one year.

LTA₄-d₅ methyl ester is supplied as a solution in hexane containing 1% triethylamine. To change the solvent, simply evaporate the hexane under a gentle stream of nitrogen and immediately add the solvent of choice. Solvents such as ethanol, DMSO, and dimethyl formamide purged with an inert gas can be used. The solubility of LTA₄-d₅ methyl ester in these solvents is approximately 50 μg/ml. The naturally occurring free acid of LTA₄-d₅ is too unstable for storage. The methyl ester is provided because of its increased stability. However, both the free acid and the methyl ester decompose rapidly under acidic conditions. To perform MS analysis of LTA₄-d₅, the methyl ester must be hydrolyzed. Alkaline hydrolysis of LTA₄-d₅ methyl ester can be performed as follows:

Prepare a hydrolysis solution consisting of degassed acetone (8 ml) and 0.25 M NaOH (2 ml) and cool it to 0°C. Evaporate the hexane solution of LTA4-d5 methyl ester just to dryness under nitrogen and immediately add 100 μl of the hydrolysis solution per 25 μg of LTA₄-d₅ methyl ester. Allow the reaction to stand under an inert atmosphere of nitrogen or argon at 22°C for 40 mintues. The resulting basic solution of LAT4-d $_5$ will be stable for about 60 minutes at room temperature or 12 hours at 0°C. Solutions used within 12 hours of hydrolysis should be discarded.

LTA₄-d₅ methyl ester is used as an internal standard for the quantification of LTA₄ methyl ester by stable isotope dilution MS. The accuracy of the sample weight in this vial is between 5% over and 2% under the amount shown on the vial. If better precision is required, the deuterated standard should be quantitated against a more precisely weighed unlabeled standard by constructing a standard curve of peak intensity ratios (deuterated versus unlabeled).

LTA₄ is synthesized in mast cells, eosinophils, and neutrophils from arachidonic acid by 5-lipoxygenase (5-LO), which or LTC₄, respectively. LTA₄, from leukocytes, is known to undergo transcellular metabolism in platelets, erythrocytes, and endothelial cells.³ Further metabolism of LTA₄ by 15-LO leads to lipoxin biosynthesis.² LTA₄ as a free acid is highly unstable. The methyl ester is stable and can be readily hydrolyzed to the free acid as needed.

- Manganaro, F., Gaudette, Y., Pombo-Gentile, A., et al. Prostaglandins 36, 859-874 (1988).
- Shimizu, T., Rådmark, O., and Samuelsson, B. Proc. Nat. Acad. Sci USA 81, 689-693 (1984).
- Samuelsson, B., Dahlén, S.-E., Lindgren, J.Å., et al. Science 237, 1171-1176 (1987).
- Maclouf, J.A. and Murphy, R.C. J. Biol. Chem. 263, 174-181 (1988).

Related Products

For a list of related products please visit: www.caymanchem.com/catalog/10006197

WARNING: This product is for laboratory research only: not for administration to humans. Not for human or veterinary DIAGNOSTIC OR THERAPEUTIC USE

MATERIAL SAFETY DATA

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