# **Product Information**



## 2,3-bis(4-hydroxyphenyl) Propionitrile

1/20 (7 7

Item No. 10008842

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1428-6/-/	
4-hydroxy-α-(4-hydroxyphenyl)-	HO, OH
benzenepropanenitrile	
DPN	
$C_{15}H_{13}NO_2$	
239.3	
≥98%	L L C
≥2 years at -20°C	но 🦳
A crystalline solid	Ν
λ <sub>max</sub> : 230, 270 nm	
	4-hydroxy-α-(4-hydroxyphenyl)- benzenepropanenitrile DPN C <sub>15</sub> H <sub>13</sub> NO <sub>2</sub> 239.3 ≥98% ≥2 years at -20°C

## Laboratory Procedures

For long term storage, we suggest that 2,3-bis(4-hydroxyphenyl) propionitrile (DPN) be stored as supplied at -20°C. It will be stable for at least two years.

DPN is supplied as a crystalline solid. A stock solution may be made by dissolving the DPN in an organic solvent purged with an inert gas. DPN is soluble in organic solvents such as ethanol, DMSO, and dimethyl formamide (DMF). The solubility of DPN in ethanol is approximately 20 mg/ml and is approximately 30 mg/ml in DMSO and DMF.

DPN is sparingly soluble in aqueous buffers. For maximum solubility in aqueous buffers, DPN should first be dissolved in DMSO and then diluted with the aqueous buffer of choice. DPN has a solubility of approximately 0.5 mg/ml in a 1:1 solution of DMSO:PBS (pH 7.2) using this method. We do not recommend storing the aqueous solution for more than one day.

DPN is an estrogen receptor  $\beta$  (ER $\beta$ ) selective agonist. DPN exhibits a 70-fold higher relative binding affinity for ER $\beta$ (18%) versus ER $\alpha$  (0.25 %), about 80-fold higher transcriptional potency (EC<sub>50</sub> = 0.85 nM versus 66 nM), and 170-fold higher relative potency (4.6% versus 0.025 %) in transcription assays.<sup>1</sup> Based on this selectivity, DPN is commonly used as a tool to evaluate the biological role of ERβ.

#### Reference

1. Meyers, M.J., Sun, J., Carlson, K.E., *et al.* Estrogen receptors- $\beta$  potency-selective ligands: Structure-activity relationship studies of diarylpropionitriles and their acetylene and polar analogus. J. Med. Chem. 44, 4230-4251 (2001).

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