# PRODUCT INFORMATION



# **Xanthine**

Item No. 31499

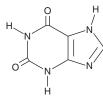
CAS Registry No.: 69-89-6

Formal Name: 3,9-dihydro-1H-purine-2,6-dione Synonyms: 2,6-Dihydroxypurine, NSC 14664

MF:  $C_5H_4N_4O_2$ FW: 152.1 **Purity:** ≥98%  $\lambda_{max}$ : 268 nm A crystalline solid UV/Vis.: Supplied as:

Storage: -20°C Stability: ≥4 years

Information represents the product specifications. Batch specific analytical results are provided on each certificate of analysis.



## **Laboratory Procedures**

Xanthine is supplied as a crystalline solid. A stock solution may be made by dissolving the xanthine in the solvent of choice, which should be purged with an inert gas. Xanthine is soluble in the organic solvent DMSO (gently warmed) at a concentration of approximately 1 mg/ml.

Further dilutions of the stock solution into aqueous buffers or isotonic saline should be made prior to performing biological experiments. Ensure that the residual amount of organic solvent is insignificant, since organic solvents may have physiological effects at low concentrations. Organic solvent-free aqueous solutions of xanthine can be prepared by directly dissolving the crystalline solid in aqueous buffers. The solubility of xanthine in PBS (gently warmed), pH 7.2, is approximately 1 mg/ml. We do not recommend storing the aqueous solution for more than one day.

## Description

Xanthine is a purine base and intermediate in the biosynthesis of uric acid (Item No. 16219).<sup>1,2</sup> It is formed during mammalian purine catabolism in the liver via oxidation of hypoxanthine (Item No. 22254) by xanthine oxidase (XO), which also oxidizes xanthine to produce uric acid. $^{1,3}$  Xanthine is also formed from guanine, xanthosine, or hypoxanthine during purine catabolism in plants.<sup>4</sup> It has been found in a variety of commercial foodstuffs, including beer yeast, mushrooms, vegetables, fish, and beef.<sup>5</sup> Urinary xanthine levels are decreased in patients with primary gout.6 Xanthine has also been used in the synthesis of xanthine derivatives that have anticancer or anti-inflammatory activities in vitro.<sup>7</sup>

#### References

- 1. Dawson, J. and Walters, M. Br. J. Clin. Pharmac. 62(6), 633-644 (2006).
- 2. Wu, S., Jia, S., and Dong, X. Int. J. Curr. Res. Chem. Pharm. Sci. 3(8), (2016).
- 3. Garcia-Gil, M., Camici, M., Allegrini, S., et al. Int. J. Mol. Sci. 19(11), 3598 (2018).
- 4. Zrenner, R., Stitt, M., Sonnewald, U., et al. Annu. Rev. Plant Biol. 57, 805-836 (2006).
- 5. Kaneko, K., Aoyagi, Y., Fukuuchi, T., et al. Biol. Pharm. Bull. 37(5), 709-721 (2014).
- 6. Puig, J.G., Mateos, F.A., Jiménez, M.L., et al. Am. J. Med. 85(4), 533-537 (1988).
- 7. Singh, N., Shreshtha, A.K., Thakur, M.S., et al. Heliyon 4(10), e00829 (2018).

WARNING
THIS PRODUCT IS FOR RESEARCH ONLY - NOT FOR HUMAN OR VETERINARY DIAGNOSTIC OR THERAPEUTIC USE.

This material should be considered hazardous until further information becomes available. Do not ingest, inhale, get in eyes, on skin, or on clothing. Wash thoroughly after handling. Before use, the user must review the complete Safety Data Sheet, which has been sent via email to your institution.

# WARRANTY AND LIMITATION OF REMEDY

Buyer agrees to purchase the material subject to Cayman's Terms and Conditions. Complete Terms and Conditions including Warranty and Limitation of Liability information can be found on our website.

Copyright Cayman Chemical Company, 12/08/2022

## **CAYMAN CHEMICAL**

1180 EAST ELLSWORTH RD ANN ARBOR, MI 48108 · USA PHONE: [800] 364-9897

[734] 971-3335

FAX: [734] 971-3640 CUSTSERV@CAYMANCHEM.COM WWW.**CAYMANCHEM**.COM